Synthesis and Evaluation of $Gly\psi(PO_2R-N)Pro$ -Containing Pseudopeptides as Novel Inhibitors of the Human Cyclophilin hCyp-18[†]

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The human cyclophilin hCyp-18, an abundant peptidyl-prolyl cis-trans isomerase (PPIase) implicated in protein folding, controls the infection of CD4+ T-cells by HIV-1, the pathologic agent of AIDS. Therefore, hCyp-18 is an interesting target for the development of novel anti-HIV-1 therapeutics. We focused on the design of transition-state analogue inhibitors of the PPIase activity of cyclophilin. Most experimental results reported in the literature suggest that hCyp-18 catalyzes the pyramidalization of the nitrogen of pyrrolidine via an H-bond network which results in the deconjugation of the amino acyl-prolyl peptide bond. We proposed the Gly ψ (PO₂R¹-N)Pro motif (R = alkyl or H) as a selective transition-state analogue inhibitor of cyclophilin. This motif was inserted in Suc-Ala-Ala-Pro-Phe-pNA, a peptide substrate of hCyp-18. The pseudopeptide Suc-Ala-Gly ψ (PO₂Et-N)Pro-Phe-*p*NA **1b** bound to hCyp-18 ($K_d = 20 \pm$ 5 μ M) and was able to selectively inhibit its PPIase activity (IC₅₀ = 15 ± 1 μ M) but not hFKBP-12, another important PPIase. Deprotection of the phosphonamidate moiety resulted in a complete lack of inhibition. We previously demonstrated that reduction of the Phe-pNA moiety caused a quantitative reduction of the affinity; however, Suc-Ala-Gly ψ (PO₂Et-N)Pro-Phe ψ - $(CH_2-NH)pNA$ **7b** still bound and inhibited hCyp-18, suggesting that the $Gly\psi(PO_2Et-N)Pro$ motif plays the major role in the binding to cyclophilin. Consequently, we propose compound **1b** as being a novel transition-state mimic inhibitor of hCyp-18.

Introduction

Immunophilins are abundant and ubiquitous peptidyl-prolyl isomerases (PPIases) which specifically catalyze the cis-trans interconversion of the amino acylprolyl amide bond.¹ As a consequence, they play a critical role in protein folding and in several related biological processes, such as cellular multiplication² and communication.³ hCyp-18, a cytosolic member of the human cyclophilin subfamily, is also the main receptor of the immunosuppressive undecapeptide cyclosporin A and, hence, is implicated in immunosuppression.⁴ Moreover, hCyp-18 is involved in the infection of T-cells by HIV-1, the pathological agent of AIDS, and, hence, participates in the control of virus multiplication.⁵⁻⁷ Therefore, the design of specific inhibitors of hCyp-18 arouses considerable interest for the development of new drugs, in particular, of anti-AIDS agents. Several strategies have been proposed for the development of cyclophilin inhibitors as novel therapeutics. These include non-immunosuppressive analogues of cyclosporin,⁸⁻¹¹ short modified peptides derived from the viral Gag polyprotein,12 nonisomerizable substrate analogues,^{13,14} and ground-state analogue inhibitors.¹⁵ Here, we present pseudopeptides mimicking the transition state¹⁶ of the PPIase activity as selective inhibitors of cyclophilin.

The mechanism of cis-trans isomerization is still unclear, but experimental data suggest that the transition state is a deconjugated amide (ketoamine form).¹⁷ The catalyzed isomerization is globally favored by the hydrophobic environment of the active site.¹⁸ The structure of the complex of hCyp18 with tetrapeptide Suc-Ala-Ala-Pro-Phe-pNA shows that Arg55 located at the catalytic subsite strongly binds the carboxyl at the P1' C-terminus. In addition, this arginine could drive the pyramidalization of nitrogen (a prerequisite to the amide hyperpolarization) by binding the pyrrolidine nitrogen at P1' via an H-bond.¹⁷ Arg55 might also assist the rotation of the acyl moiety by a transient interaction (Scheme 1, A and A').¹⁹⁻²³ This hypothesis is supported by results from Janda and co-workers²⁴ and Schultz and collaborators²⁵ who generated PPIase catalytic antibodies using a ketoamide-containing hapten. In the latter work, a ketoamide-containing transition-state analogue inhibitor was developed. This compound, however, is not selective and inhibits both hCyp-18 and hFKBP-12, another important PPIase.^{25,26} Nevertheless, these data indicate that by mimicking a rotating amide bond, novel transition-state analogue inhibitors of hCyp-18 might be able to be designed. For this pupose, it is important to study the differences found in the various PPIase subfamilies. The hypothesis of a tetrahedral covalent intermediate has been ruled out for the cis-trans isomerization process catalyzed by hCyp-18, while the crystallographic structure of a complex between a dipeptide and Pin1, a human parvulin implicated in the control of mitosis²⁷ and involved in neurodegeneration,²⁸ suggests the existence of a tetrahedralized carbonyl carbon at P1.29 Moreover, Pin1 possesses several nucleophilic residues that might participate in the catalysis.²⁹ On the other hand, FKBP does not possess a polar

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Glyw (PO2R-N)Pro-Containing Pseudopeptides

Scheme 1. Proposed Mechanism of the hCyp-18-Catalyzed Isomerization: (A) Arg55-Assisted Deconjugation of the Amide via an H-Bond with the Guanidinium Side Chain and Pyramidalization of the Proline Nitrogen, (A') Arg55-Assisted Rotation of the Acyl Group via an H-Bond with the Guanidinium Side Chain, (B) the Phosphonamide Transition-State Analogue Which Might Mimic both Putative Transition States A and A', and (C) Ketoamide Transition-State Analogue Which Is Supposed To Mimic the Rotating Acyl Moiety



residue at the active site and interacts with its substrate in a mode that is completely different from that of hCyp-18 and Pin1. 19

Phosphinic or phosphonamide amino acyl-proline surrogates might mimic the rotating amide bond. We recently reported the synthesis and biochemical evaluation of the phosphinic pseudopeptides Ac-Ala ψ (PO₂R-CH)Pro-*p*CMA (R = H or CH₃). Though the phosphinic moiety is chemically stable and resistant to proteolysis, these peptides are not able to inhibit the PPIase activity of hCyp-18.³⁰ We investigated the potential of phosphonamide surrogates of Gly-Pro as novel transition-state mimics of PPIase activity. Though these surrogates are less chemically resistant than their phosphinic equivalent,³¹ the phosphonamide nitrogen is slightly pyramidalized,³² and therefore, its doublet is available for establishing a strong H-bond with the guanidinium moiety of Arg55 (Scheme 1, B).

We report herein on the synthesis and biochemical evaluation of phosphonamide-containing pseudopeptides as transition-state analogue inhibitors of hCyp-18. The corresponding ketoamide was prepared as a reference for the test tube assays (Scheme 1, C).

Experimental Section

Abbreviations. Ala ψ (COCO-N)Pro, ((*S*)-2-oxo-3-aminobutanoyl)proline; DCC, dicyclohexylcarbodiimide; DIPEA, diisopropylethylamine; Gly ψ (COCO-NH)Pro, (2-oxo-3-aminopropanoyl)proline; Gly ψ (PO₂R-NH)Pro, ethyl aminomethylphosphonylproline (R = Et) or aminomethylphosphonylproline sodium salt (R = Na); HOBT, 1-hydroxybenzotriazole; *p*CMA, 4-carboxymethylaniline; Phe ψ (CH₂-NH)*p*NA, 1-(4-nitrophenyl)amino-(*R*)-2-benzyl-2-aminoethane; *p*NA, 4-nitroaniline; SD, standard deviation; Suc, succinyl; THF, tetrahydrofuran; Val ψ (COCO-NH)Pro, ((*S*)-2-oxo-3-amino-4-methylpentanoyl)-proline.

All the reagents employed were of analytical grade and were purchased from Aldrich Chemical Co. or Sigma. Amino acids and coupling reagents were obtained from Novabiochem, Bachem, or Sigma. Suc-Ala-Gly-Pro-Phe-pNA, Suc-Ala-Ala-Pro-Phe-pNA, Suc-Ala-Ala-Pro-Arg-pNA, and Suc-Ala-Leu-Pro-Phe-pNA were purchased from Bachem. Suc-Ala-Ala-Pro-Phe-DFA was synthesized as previously reported.³⁴ All other solvents were of analytical grade and were used without further purification. All intermediates were purified by flash chromatography on 40–60 μ m (230–400 mesh) Merck silica gel. Peptides were purified by RP-HPLC. All compounds were characterized by ¹Ĥ, ¹³C, and ³¹P NMR (recorded on a Bruker AVANCE 250 NMR spectrometer); δ values are given in parts per million (ppm), and J values are in hertz. Electrospray mass spectra (Atheris Laboratories, Geneva, Switzerland) were recorded on a Micromass Platform II (Micromass, Altrincham, U.K.). Fluorimetric determination of K_d was done using a Jasco FP-750 spectrofluorometer equipped with a 250 μ L thermostated cell. Kinetic assays were performed using a Kontron Uvikon 930 spectrophotometer equipped with a 2 mL thermostated cell. The recombinant hCyp-18 was obtained as previously reported.12

Pht=GlyΨ(PO₂Et-N)Pro-Phe-pNA (3a and 3b). Compound 2 (2.97 g, 10 mmol) was refluxed with phosphorus pentachloride (2.41 g, 1.1 equiv) in freshly dried benzene (15 mL) for 7 h under argon. After removal of the solvent under reduced pressure, the crude product dissolved in freshly dried THF (10 mL) was added dropwise to a solution of Pro-PhepNA (0.5 equiv) and triethylamine (3.48 mL, 2.5 mmol) in freshly dried THF. The mixture was stirred overnight under argon at room temperature. After evaporation of the solvent, the crude product was dissolved in dichloromethane and washed twice with 10% citric acid, saturated sodium bicarbonate, and brine. The organic layer was dried over sodium sulfate, and the volume was reduced in vacuo. Purification by silica gel flash chromatography (eluent, 65:35 ethyl acetate: chexane ratio) gave 3a (939 mg, 1.48 mmol) and 3b (576 mg, 0.91 mmol) in 48% overall yield.

3a: ¹H NMR (CDCl₃) δ 7.84 (d, J = 5.1, 2H) + 7.63 (d, J = 7.1, 2H) + 7.42–7.17 (m, 12H), 5.05–4.94 (m, 1H), 4.33–4.02 (m, 5H), 3.81 (dd, J = 14.1, J' = 4.6, 1H), 3.53–3.45 (bm, 1H), 3.29–3.15 (m, 1H), 3.10 (dd, J = 14.1, J' = 12.3), 2.09–2.04 + 1.80–1.75 + 1.62–1.51 + 1.30–1.22 (4m, 4H), 1.42 (t, J = 7.0, 3H); ¹³C NMR (CDCl₃) δ 172.7 + 170.5 + 168.0, 144.1 + 142.8 + 137.8 + 134.4 + 130.9 + 129.1 + 128.5 + 126.7 + 124.1 + 123.3 + 118.9, 62.7 (d, $J'_{C-P} = 7.2$), 61.4 (d, $J'_{C-P} = 1.1$), 53.5, 47.5 (d, $J'_{C-P} = 2$), 36.8, 33.7 (d, $J'_{C-P} = 141.3$), 31.3 (d, $J'_{C-P} = 7.5$), 25.6 (d, $J'_{C-P} = 7.2$), 16.4 (d, $J'_{C-P} = 6.0$); ³¹P NMR (CDCl₃, decoupled) δ 25.8; DCI-MS (NH₃) m/z 651 (MNH₄⁺, 9%), 634 (MH⁺, 100%); combustion analysis (C₃₁H₃₂N₅O₈) C, H, N.

3b: ¹H NMR (CDCl₃) δ 8.09 (d, J = 9.25, 2H) + 7.96 (d, J = 9.25, 2H), 7.88–7.85 + 7.78–7.75 (2m, 4H), 7.29–7.25 + 7.18–7.14 (5H), 5.01–4.92 (m, 1H), 4.21–4.07 (m, 5H), 3.83–3.71 + 3.66–3.46 + 3.23–3.04 (3m, 4H), 2.41–2.20 + 2.09–1.81 + 1.74–1.51 (3m + 4H), 1.19 (t, J = 7, 3H); ¹³C NMR (CDCl₃) δ 172.4 + 170.4 + 167.2, 144.5 + 143.3 + 136.3 + 134.6 + 131.6 + 129.4 + 128.9 + 127.3 + 124.5 + 123.8 + 119.9, 61.7 (d, $\mathcal{P}_{C-P} = 6.2$), 61.4 (d, $\mathcal{P}_{C-P} = 6.5$), 53.4, 48.6 (d, $\mathcal{P}_{C-P} = 2.6$), 37.2, 32.5 (d, $\mathcal{J}_{C-P} = 137.9$), 31.7 (d, $\mathcal{P}_{C-P} = 7.7$), 25.2 (d, $\mathcal{J}_{C-P} = 7.5$), 16.3 (d, $\mathcal{J}_{C-P} = 6.8$); ³¹P NMR (CDCl₃, decoupled) δ 24.2; DCI-MS (NH₃) *m*/*z* 651 (MNH₄⁺, 14%), 634 (MH⁺, 100%); combustion analysis (C₃₁H₃₂N₅O₈) C, H, N.

Fmoc-Ala-Gly Ψ (**PO**₂**Et-N**)**Pro-Phe**-*p***NA** (4a and 4b). Compounds **3a** (0.63 g, 1 mmol) and **3b** (0.5 g, 0.79 mmol) were treated with 1.0 M hydrazine in THF (4.0 equiv) overnight at room temperature. The mixture was filtered, and the cake was washed with THF (3 times). The crude product was dissolved in dry dichloromethane (10 mL) and treated at 0 °C with Fmocalanine (9 equiv), DCC (9.9 equiv), HOBT (9 equiv), and DIPEA (18 equiv). The mixture was stirred overnight at room temperature. The crude mixture was filtered and washed with 10% citric acid (2 times), saturated sodium bicarbonate, and brine. The organic layer was dried over sodium sulfate. Purification by silica gel flash chromatography (eluent, 65:35 ethyl acetate: chexane ratio) gave **4a** (690 mg, 88%) and **4b** (580 mg, 92%).

4a: ¹H NMR (CDCl₃) δ 8.10 (d, J = 7.5, 2H) + 7.88 (d, J = 9.0, 2H) + 7.75 (d, J = 7.3, 2H) + 7.54 (d, J = 7.2, 2H) + 7.42– 6.80 (bm, 9H), 5.48–5.45 (m, 1H), 4.97–4.93 (m, 1H), 4.41– 4.30 + 4.20–3.92 (2bm, 8H), 3.47–3.44 + 3.22–3.07 (2bm, 4H), 1.97–1.80 + 1.75–1.53 (2b, 4H), 1.38–1.20 (bm, 6H); ¹³C NMR (CDCl₃) δ 173.8 + 173.5 + 173.4 + 171.3, 156.9 + 156.2, 144.3–118.8 (complex), 67.3, 62.1, 61.4 (d, $J^{2}_{C-P} = 6.6)$, 54.9, 53.7 + 47.0, 50.4 + 49.1 + 46.7, 37.2, 36.3 (d, $J^{1}_{C-P} = 147.3)$, 31.4 (d, $J^{3}_{C-P} = 7.2$), 25.8 (d, $J^{3}_{C-P} = 7.3$), 18.0, 16.5 (d, $J^{3}_{C-P} = 5.9$); ³¹P NMR (CDCl₃, decoupled) δ 26.5 (minor) + 27.1 (major); DCI-MS (NH₃) m/z 797 (MH⁺, 30%), 415 ([Pro-Phe*p*NA]H⁺, 35%), 383 ([Fmoc-Ala-CH₂PO₂Et]H⁺, 100%).

4b: ¹H NMR (CDCl₃) δ 8.08 (d, J = 9.3), 7.91 (d, J = 9.3), 7.78–7.64 + 7.57–7.50 + 7.46–7.11 (bm, 13H), 5.40–5.31 (m, 1H), 4.97–4.87 (m, 1H), 4.49–3.96 (2bm, 8H), 3.61–3.32 + 3.11–2.98 (2bm, 4H); ¹³C NMR (CDCl₃) δ 173.4 + 172.9 + 170.3, 157.4 + 156.1 + 155.7, 144.2–110.8 (complex), 67.2, 63.8 (d, $J^2_{C-P} = 6.2$), 62.0 (d, $J^2_{C-P} = 4.5$) + 61.0 (d, $J^2_{C-P} = 6.9$), 53.9, 50.9 + 47.7 + 47.0 + 47.0 (d, $J^2_{C-P} = 9.2$), 36.9, 34.1 (d, $J^1_{C-P} = 141.8$), 31.6 (d, $J^2_{C-P} = 6.9$), 25.5 (d, $J^2_{C-P} = 8.2$), 18.1, 16.2 (d, $J^3_{C-P} = 6.8$); ³¹P NMR (CDCl₃, decoupled) δ 27.6; DCI-MS (NH₃) *m*/*z* 797 (MH⁺, 13%), 415 ([Pro-Phe-*p*NA]H⁺, 29%), 383 ([Fmoc-Ala-CH₂PO₂Et]H⁺, 100%).

Suc-Ala-GlyΨ(PO₂Et-N)Pro-Phe-*p***NA (1a and 1b).** Compounds **4a** (500 mg, 0.79 mmol) and **4b** (450 mg, 0.58 mmol) were treated with 20% diisopropylamine in DMF for 3 h at room temperature. After evaporation of the solvent under reduced pressure, methanol was chased over the product (3 times). The crude product was treated with succinic anhydride (2.0 equiv) and trietylamine (3.0 equiv) overnight at room temperature. After evaporation of the solvent under reduced pressure, the crude product was purified by RP-HPLC (semi-preparative column Vydac C₁₈; rate, 4 mL min⁻¹; linear gradient, acetonitrile:acetic acid (0.25% in water) 25:75 to 75: 25 in 30 min).

1a (50 mg, 12%): $t_{\rm R}$ 11.09 min (95%); ¹H NMR (CD₃OD) δ 8.19 (d, J = 9.0) + 7.84 (d, J = 9.2), 7.36–7.16 (bm, 5H), 4.90– 4.82 (m, 1H), 4.39 (q, J = 7.2, 1H), 4.28–4.17 (m, 1H), 4.06– 3.90 (m, 2H), 3.83–3.69 (m, 2H), 3.30–3.10 (bm, 4H), 2.68– 2.55 + 2.55–2.41 (bm, 4H), 2.19–2.03 + 1.84–1.52 (2bm, 4H), 1.35 (d, J = 7.2, 3H), 1.26 (t, J = 7.0, 3H); ¹³C NMR (CD₃OD) δ 176.4 + 175.3 + 175.2 + 174.6 + 172.4, 145.7 + 144.7 + 138.5 + 130.5 + 129.5 + 127.9 + 125.7 + 120.7, 62.7 (d, $J^{2}_{\rm C-P}$ = 142.7), 32.7 (d, $J^{2}_{\rm C-P}$ = 6.9), 31.4 + 30.2, 26.6 (d, $J^{3}_{\rm C-P}$ = 6.5), 17.7, 16.6 (d, $J^{2}_{\rm C-P}$ = 6.4); ³¹P NMR (CD₃OD, decoupled) δ 25.7; ES-MS (negative ionization) m/z 674.19.

1b (50 mg, 12%): $t_{\rm R}$ 12.45 min (92%); ¹H NMR (CD₃OD) δ 8.19 (d, J = 9.15, 2H) + 7.87 (d, J = 9.15, 2H), 7.33–7.17 (bm, 5H), 4.81–4.73 (m, 2H), 4.34 (q, J = 7.2, 1H), 4.15–3.96 (m, 4H), 3.70–3.56 + 3.26–3.13 (2m, 4H), 2.64–2.47 (bm, 4H), 2.21–2.02 + 1.91–1.69 (2bm, 4H), 1.35 (d, J = 7.5, 3H), 1.24 (t, J = 7.05, 3H); ¹³C NMR (CD₃OD) δ 176.3 + 175.2 + 175.1 + 174.7 + 172.5, 145.7 + 144.8 + 138.1 + 130.4 + 129.6 + 128.0 + 125.7 + 125.6 + 120.8 (complex), 62.6–62.3 (m), 56.5, 50.2 (d, $\mathcal{J}_{\rm C-P} = 7.3$), 38.4, 35.6 (d, $\mathcal{J}_{\rm C-P} = 144.9$), 32.8 (d, $\mathcal{J}_{\rm C-P} = 7.0$), 31.4 + 30.3, 26.3 (d, $\mathcal{J}_{\rm C-P} = 7.5$), 17.7, 16.6 (d, $\mathcal{J}_{\rm C-P} = 5.8$); ³¹P NMR (CD₃OD, decoupled) δ 25.8; ES-MS (negative ionization) m/z 674.30.

Ac-Ala-Gly Ψ (PO₂Et-N)Pro-PheY(CH₂-NH)*p*NA (7a and 7b). Compounds **6a** (210 mg, 0.26 mmol) and **6b** (208 mg, 0.26 mmol) were treated with 20% diisopropylamine in DMF for 3 h at room temperature. After evaporation of the solvent under reduced pressure, methanol was chased over the product (3 times). The crude product was treated with acetyl chloride (2.0 equiv) and DIPEA (3.0 equiv) overnight at room temperature. After evaporation of the solvent under reduced pressure, the crude product was purified by preparative thin-layer chromatography (eluent, chloroform:methanol with 98:2, then 97:3, and finally 95:5 ratios) or by RP-HPLC (semipreparative column Vydac C₁₈; rate, 4 mL min⁻¹; linear gradient, aceto-

nitrile:acetic acid (0.25% in water) 25:75 to 75:25 in 30 min) to give **7a** (121 mg, 77%) and **7b** (128 mg, 82%).

7a: $t_{\rm R}$ 12.59 (>95%); ¹H NMR (CDCl₃) δ 8.02 (d, J = 9.2, 2H), 7.51 (bd, J = 7.95, 1H), 7.33-7.13 (m, 5H), 6.50 (d, J = 9.2, 2H), 6.30 (bd, J = 7.1, 1H), 6.19 (bt, J = 5.6, 1H), 4.44 (t, J = 7.0, 1H), 4.40-4.31 (m, 1H), 4.02-3.88 (m, 5H), 3.41-3.25 + 3.10 - 3.01 + 2.97 - 2.93 (3m, 6H), 1.99 (s, 3H), 2.04-1.94 + 1.81 - 1.62 (2m, 4H), 1.36 (d, J = 7.0, 3H), 1.26 (t, J = 7.0, 3H); ¹³C NMR (CDCl₃) δ 173.8 + 172.7 + 172.6 + 170.5, 137.4 + 129.1 + 128.6 + 128.5 + 127.7 + 126.8 + 126.4 + 126.3 + 110.9, 61.6 (d, $\mathcal{P}_{\rm C-P} = 4.15$), 61.2 (d, $\mathcal{P}_{\rm C-P} = 6.8$), 50.3 + 49.1, 47.5 (d, $\mathcal{P}_{\rm C-P} = 3.6$), 46.2, 38.1, 34.8 (d, $\mathcal{I}_{\rm C-P} = 143.3$), 31.1 (d, $\mathcal{P}_{\rm C-P} = 6.5$), 25.1 (d, $\mathcal{P}_{\rm C-P} = 6.9$), 23.0, 17.9, 16.3 (d, $\mathcal{P}_{\rm C-P} = 6.3$); ³¹P NMR (CDCl₃, decoupled) δ 27.1; ES-MS 602.4.

7b: $t_{\rm R}$ 11.90 (>95%); ¹H NMR (CDCl₃) δ 8.00 (d, J = 10.0, 2H), 7.55–7.46 (b, 1H), 7.34–7.15 (m, 5H), 6.51 (d, J = 10.0, 2H), 6.44 (d, J = 7.5, 1H), 6.16–6.08 (b, 1H), 4.50–4.29 (m, 2H), 4.23–3.84 (m, 5H, Ha), 3.68–3.55 + 3.43–3.22 + 3.18–2.97 (3m, 6H), 2.04 (minor) + 2.02 (major) (s, 3H), 2.11–1.99 + 1.81–1.62 (2m, 4H), 1.34 (d, J = 7.5, major) + 1.30 (d, J = 5.0, minor), 1.26 (t, J = 5.0, major) + 1.25 (t, J = 5.0, minor); ¹³C NMR (CDCl₃) δ 174.5 + 173.0 + 172.8 + 170.4, 137.6 + 137.4 + 129.1 + 128.5 + 126.7 + 110.9, 61.8 (d, $J^2_{\rm C-P} = 6.8$), 60.9 (d, $J^2_{\rm C-P} = 3.6$), 50.2 + 49.0, 47.3 (d, $J^2_{\rm C-P} = 3.4$), 46.4, 38.1, 36.2 (d, $J^1_{\rm C-P} = 146.0$), 31.5 (d, $J^3_{\rm C-P} = 7.1$), 25.7 (d, $J^8_{\rm C-P} = 6.8$), 23.1, 17.9, 16.3 (d, $J^3_{\rm C-P} = 6.2$); ³¹P NMR (CDCl₃, decoupled) δ 26.7; ES-MS m/z 602.3.

Ac-Ala-GlyΨ(PO₂Bn-N)Pro-Phe-pCMA (11). Compound **10** (85 mg, 87 μ mol) was treated with 20% diisopropylamine in DMF for 3 h at room temperature. After evaporation of the solvent under reduced pressure, methanol was chased over the product (3 times). The crude product was treated with acetyl chloride (13 mL, 2.0 equiv) and DIPEA (45 mL, 3.0 equiv) overnight at room temperature. After evaporation of the solvent under reduced pressure, the crude product was purified by preparative thin-layer chromatography (eluent, 94:6 chloroform:methanol ratio) to give 11 (31 mg, 52%): ¹H NMR $(CDCl_3) \delta$ 7.98 (d, J = 10.0, 2H) + 7.84 (d, J = 10.0, 2H), 7.68 (d, J = 7.5, 1H) + 7.37-7.18 (m, 9H), 5.96 (d, J = 7.3, 1H), 5.00 (AB, $\delta_A = 5.07$, $\delta_B = 4.94$, $J_{AB} = 11.8$) + 5.00 (AB, $\delta_A =$ 5.03, $\delta_{\rm B} =$ 4.97, $J_{\rm AB} =$ 11.7) (m, 2H), 4.85–4.96 + 4.42–4.34 + 4.08-4.05 + 3.94-3.84 (4m, 5H), 3.87 (s, 3H), 3.51-2.95 (m, 4H), 1.79 (s, 3H), 1.80-1.73 + 1.54-1.48 (2m, 4H), 1.26(d, J = 7.1, 3H); ¹³C NMR (CDCl₃) δ 173.7 + 173.2 + 171.0 + 170.7 + 166.6, 142.6 + 137.7 + 136.0 + 135.9 + 130.6 + 129.3+ 128.8 + 128.4 + 128.0 + 126.7 + 125.4 + 119.3, 66.7 (d, $J^{2}_{C-P} = 7.0$), 61.3 (d, $J^{2}_{C-P} = 2.3$), 54.9 + 52.0 + 48.6 + 47.0, 36.6, 36.6 (d, $\mathcal{J}^{1}_{C-P} = 146.0$), 31.1 (d, $\mathcal{J}^{3}_{C-P} = 7.0$), 25.2 (d, \mathcal{J}^{3}_{C-P} = 7.0), 22.8, 16.9; ³¹P NMR (CDCl₃, decoupled) δ 28.1; ES-MS m/z 691.5.

Ac-Ala-GlyΨ(PO₂H-N)Pro-Phe-pCMA (12). Compound 11 (23 mg, 33 μ mol) dissolved in a 50:50 methanol/water mixture (7 mL) was treated with sodium bicarbonate (8 mg, 2.8 equiv) and 10% Pd-C (11 mg) under 4 bar of hydrogen for 2 h. The catalyst was removed by filtration, and the solvent was evaporated under reduced pressure. The crude product was dissolved in a 50:50 water/acetonitrile mixture and freezedried. Compound 12 was used without further purification: ¹H NMR (CD₃OD) δ 8.01–7.92 (m, 4H), 7.32–7.19 (m, 5H), 4.90-4.78 (m), 4.32 (q, J = 7.0, 1H), 4.00-3.93 (m, 1H), 3.88(s, 3H), 3.64 (dd, J = 15.2, J = 11.8, 1H) + 3.52 (dd, J = 13.9, J' = 4.0, 1H) + 3.30 - 3.19 (m, 1H) + 3.01 - 2.91 (m, 1H), 1.96(s, 3H), 1.90-1.75 + 1.65-1.50 (2m, 4H), 1.29 (d, J = 5.0, 3H); 13 C NMR (CD₃OD) δ 178.5 + 174.4 + 174.3 + 173.1 + 173.0 + 168.3, 144.3 + 139.1 + 131.1 + 130.2 + 129.5 + 127.7 + 126.6+ 121.7, 63.6, 56.3 + 52.5 + 50.5, 38.3, 37.3 (d, $J^{1}_{C-P} = 128.9$), 32.6, 26.3 (d, $J^{3}_{C-P} = 5.5$), 22.5, 17.9; ³¹P NMR (CD₃OD, decoupled) δ 16.4; ES-MS (negative ionization) m/z 601.4.

Biochemical Evaluation. (1) Determination of K_{d} . Fluorimetric titration was carried out as previously reported.^{12,33}

(2) Enzymatic Assays. The PPIase uncoupled assay was carried out in a 35 mM Hepes buffer (pH 7.8) at 10 ± 0.5 °C using Suc-Ala-Ala-Pro-Phe-DFA as a substrate.³⁴ The PPIase

trypsin-hCyp-18-coupled assay was carried out in a 35 mM Hepes buffer (pH 8.6) at 10 \pm 0.5 °C using Suc-Ala-Ala-Pro-Arg-pNA, $^{3.35}$ and the α -chymotrypsin-FKBP-coupled assay was carried out in a 35 mM Hepes buffer (pH 7.8) at 10 \pm 0.5 °C using Suc-Ala-Leu-Pro-Phe-pNA as a substrate. $^{21.36}$ Data were processed according to the literature. 12

Results and Discussion

Design Rationale. Transition-state analogues were designed on the basis of information obtained from crystallographic data and structure-activity relationship studies. We chose a substrate of immunophilins, the tetrapeptide Suc-Ala-Ala-Pro-Phe-pNA,³⁶ as a template for the development of novel hCyp-18 inhibitors. Several structures of hCyp-18-peptide complexes have been resolved and the main interactions delineated.¹⁷ Structure-activity relationship studies also provided interesting information; we recently demonstrated³⁷ the existence of two functionally independent subsites: (i) the S1' subsite which contains the catalytic residue and controls the selectivity for proline at P1' and (ii) the S2'-S3' subsite which specifically recognizes the Xaa-pNA moiety or Xaa-pNA surrogates.^{35,37} Therefore, the P1' proline is dispensable for the binding of a ligand by hCyp-18 provided the peptide possesses either an XaapNA or an Xaa-pNA analogue at P2'-P3'.³⁷ However, pNA-containing peptides without either a proline residue or a proline mimic cannot inhibit hCyp-18.^{37,38} The N-terminus of the tetrapeptide does not participate in the binding because it protrudes outside the catalytic site. The binding is mainly controlled by the carboxyl at P2' which might establish a strong H-bond with Trp121.³⁷ The residues at P1, P1', and P2' must be in the L-configuration.³⁵ Cyclophilins prefer Ala, Val, Leu, Glu, Phe, or Gly at P1, whereas FKBPs recognize preferentially Phe-Pro or Leu-Pro. In particular, the selectivity factor for hCyp-18 versus hFKBP-12 is 3500 and 1000 for Glu and Gly, respectively, but only 60 for Ala at P1'.²¹ Though Glu is preferred at P1, hCyp-18 is indeed quite selective of the Gly-Pro moiety. The introduction of the Gly ψ (PO₂Et-NH)Pro motif generates only one novel stereogenic center and, hence, provides only two diastereomers, whereas the Glu analogue would give a mixture of four diastereomers. Therefore, we designed the tetrapeptide analogue 1 (mixture of both diastereomers 1a and 1b) which contains a phosphonamide isostere of Gly-Pro.32

Synthesis of Pseudopeptides. Phosphonamidatecontaining pseudopeptides have been widely used in medicinal chemistry. However, to our knowledge, there is only one example of synthesis of phosphonamide isosteres of the Xaa-Pro motif reported in the literature.³⁹ Diethyl phthalimidomethylphosphonate **2** was prepared according to Yamauchi et al.⁴⁰ Activation with phosphorus pentachloride in refluxing benzene, as previously described,⁴⁰ and coupling to Pro-Phe-pNA yielded the diastereomeric mixture of 3a and 3b which were separated by silica gel flash chromatography (Scheme 2). Hydrazinolysis of the phthalimide and subsequent coupling with Fmoc-alanine using DCC and HOBT gave the corresponding pseudopeptides 4a and 4b in good yields. Removal of the Fmoc group and succinvlation led to the target compounds 1a and 1b which were purified by RP-HPLC (Scheme 2). The corresponding phosphonamidate could not be obtained

Scheme 2. Synthesis of

Suc-Ala-Gly ψ (PO₂Et-N)Pro-Phe-*p*NA **1a/b**,

Ac-Ala-Gly ψ (COCO-N)Pro-Phe ψ (CH₂-NH)*p*NA **7a/b**, and Ac-Ala-Gly ψ (PO₂Na-N)Pro-Phe-*p*CMA **12**



Table 1. Synthesis of Suc-Ala-Gly ψ (PO₂Et-N)Pro-Phe-*p*NA**1a/b**, Ac-Ala-Gly ψ (COCO-N)Pro-Phe ψ (CH₂-NH)*p*NA**7a/b**, andAc-Ala-Gly ψ (PO₂Na-N)Pro-Phe-*p*CMA**12**

\mathbb{R}^1	\mathbb{R}^2	R ³	compounds
Et	Phe- <i>p</i> NA	Suc	1a/1b
Et		-	2
Et	Phe- <i>p</i> NA	-	3a/3b
Et	Phe-pNA	_	4a/4b
Et	$Phe\hat{\psi}(CH_2-NH)pNA$	_	5a/5b
Et	$Phe\psi(CH_2-NH)pNA$	_	6a/6b
Et	$Phe\psi(CH_2-NH)pNA$	Ac	7a/7b
Bn		_	8
Bn	Phe-pCMA	_	9
Bn	Phe-pCMA	_	10
Bn	Phe-pCMA	Ac	11
Η	Phe-pCMA	Ac	12

by saponification of the ethyl esters **1a** and **1b**. This might be due to the sensitivity of the Phe-*p*NA moiety to the harsh basic conditions employed to hydrolyze phosphonamidic esters. Therefore, compounds 7a and **7b** were prepared by a similar route starting from Phe ψ -(CH₂-NH)*p*NA, a reduced equivalent of Phe-*p*NA which has an enhanced chemical stability.²⁵ The peptides were capped with an acetyl group in order to avoid a possible intramolecular hydrolysis of the acid-sensitive phosphonamide bond³⁹ as observed with compounds **1a** and **1b**. However, we did not succeed in deprotecting the ethyl ester of pseudopeptides 7a and 7b. Finally, we protected the phosphonamidate moiety with a benzyl ester which can be removed by hydrogenolysis.⁴¹ This strategy implied a replacement of the *p*NA C-terminus with a *p*CMA equivalent which is resistant to hydrogenolysis and does not affect the affinity for hCvp-18.37 The phosphonamidate 11 was obtained from dibenzyl ester 8 as depicted in Scheme 2. Due to the low chemical stability of protonated phosphonamidates, the benzyl phosphonamidate was selectively deprotected by hydro-

Table 2. Biochemical Evaluation of Pseudopeptides 1a/b, 7a/b, 12, and 13

compounds	$K_{\rm d} \pm { m SD} \; (\mu { m M})^a$	$\mathrm{IC}_{50}\pm\mathrm{SD}\;(\mu\mathrm{M})^b$
Suc-Ala-Gly-Pro-Phe-pNA	$\begin{array}{c} 140\pm10\\ 210\pm100 \end{array}$	1450 ± 60 5400 ± 400
Suc-Ala-Gly ψ (PO ₂ Et-N)Pro-Phe- <i>p</i> NA 1b	$\begin{array}{c} 210 \pm 100 \\ 20 \pm 5 \end{array}$	$\begin{array}{c} 5400 \pm 400 \\ 15 \pm 1 \end{array}$
Ac-Ala-Gly ψ (PO ₂ Et-N)Pro-Phe ψ (CH ₂ -NH) p NA 7a Ac-Ala-Gly ψ (PO ₂ Et-N)Pro-Phe ψ (CH ₂ -NH) p NA 7b	1300 ± 200 200 ± 15	no inhibition at 500 μ M ^c 30% inhibition at 500 μ M ^c
Ac-Ala-Gly ψ (PO ₂ ⁻ -N)Pro-Phe- <i>p</i> CMA 12	$\begin{array}{c} 200 \pm 13 \\ 79 \pm 4 \end{array}$	no inhibition at 100 μ M ^d
Ac-Ala-Gly ψ (COCO-N)Pro-Phe- <i>p</i> NA 13 Suc-Ala-Ala-Pro-Phe- <i>p</i> NA	127 ± 7 135 ± 20	$215 \pm 40 \\ 540 \pm 70$
Ac-Ala-Ala-Pro-Phe- <i>p</i> NA	$\frac{133 \pm 20}{145 \pm 15}$	$\begin{array}{c} 540 \pm 70 \\ 640 \pm 120 \end{array}$

^{*a*} K_d values were deduced from the fluorescence titration curves of the hCyp-18 Trp121 upon addition of inhibitors^{12,33} (n = 2). ^{*b*} IC₅₀ values were obtained using the standard trypsin-coupled PPIase assay^{13,35,36} at 10 °C (n = 2). ^{*c*} Due to the significant absorbance at 390 nm of Phe ψ (CH₂-NH)-*p*NA-containing peptides **7a/b**, the IC₅₀ values were evaluated using the standard uncoupled PPIase assay³⁴ at 10 °C; the maximum inhibitor concentration employed was 0.5 mM due to residual absorbance at 246 nm. ^{*d*} Because of its low solubility, the maximum inhibitor concentration employed was 0.1 mM.

genolysis in the presence of sodium bicarbonate, and compound **12** was isolated as a sodium salt and used without further purification.⁴¹ The ketoamide analogue **13** was synthesized starting from racemic 3-aminolactate as previously reported.¹²

Biochemical Evaluation of Compounds 1a/b, 7a/ b, **12**, **and 13**. We first checked that the Ala to Gly change at P1 did not significantly modify the affinity. For this purpose, Suc-Ala-Gly-Pro-Phe-pNA was tested as a ligand of hCyp-18 by fluorescence titration.^{12,33,37} As expected, the $K_{\rm d}$ was similar to those observed for other Phe-pNA-terminated peptides. We have previously demonstrated that efficient binding at the S2'-S3' subsite of cyclophilin does not imply that the compound is able to inhibit the PPIase activity occurring at the S1' subsite.³⁷ Consequently, we tested the ability of the peptides to inhibit the PPIase activity of hCyp-18 using the standard trypsin-coupled assay with Suc-Ala-Ala-Pro-Arg-*p*NA as a substrate.¹³ The Ala to Gly substitution caused a 3-fold increase in the IC₅₀ value (Table 2). This might be related to the particular behavior of the Gly-Pro moiety.²⁰

Pseudopeptides 1a/b, 7a/b, 12, and 13 were tested as ligands of hCyp-18. The results, summarized in Table 2, show that peptides 1a and 13 bind the enzyme with an affinity equivalent to that of the model substrate. Pseudopeptide **1b** is a good ligand of hCyp-18 ($K_d = 20$ μ M). Evaluation of the IC₅₀ value demonstrated that compound **1b** is a good inhibitor of cyclophilin ($IC_{50} =$ $15 \pm 1 \ \mu$ M): Replacement of the amide with an ethyl phosphonamidate surrogate resulted in an improvement in the IC₅₀ value by 2 orders of magnitude. Timedependent inhibition experiments demonstrated that compound **1b** is not a slow-binding inhibitor of cyclophilin because the IC_{50} value does not significantly change after 2 h of preincubation. In contrast, diastereomer **1a** is a poor inhibitor of the PPIase activity. Absolute configuration of the phosphorus stereogenic center seems to be critical for the interaction, suggesting that the ethyl phosphonamidate moiety tightly interacts at the S1' subsite.

Compound **1b** at concentrations up to 1 mM did not inhibit the human FK506-binding protein hFKBP-12. Although compound **1b** is slightly less potent than the ketoamide-containing peptide developed by Schultz and co-workers,²⁵ it selectively inhibits hCyp-18.

We have also investigated the influence of a charged phosphonamidate isostere (peptide **12**) on the biochemical activity. Though peptide **12** binds cyclophilin, no inhibition of the PPIase activity was observed at 100 μ M. This is probably due to the introduction of a negative charge because N-terminal modification does not affect the biological properties of the peptides.³⁵ Therefore, compound **12** is likely to interact with hCyp-18 via the Phe-*p*CMA moiety.³⁷

An important question to be addressed was the real influence of the phosphonamide motif on the potency of the inhibitor. Therefore, compounds 7a and 7b bearing a reduced amide at P2' were tested as well. Reduction of the C-terminal amide has been shown to dramatically affect both the affinity and inhibitory potency of model peptides.³⁷ Not surprisingly, we noticed that the reduction of the important Phe-pNA moiety caused a fall in affinity by 1 order of magnitude relative to that of the corresponding amide. However, **7b**, the reduced equivalent of **1b**, still inhibited hCyp-18. The reduction of the amide resulted in a 50-fold increase in the IC₅₀ value for the most active compound **7b** versus 1b; this probably reflects the lack of interaction of the compound at the S2'-S3' subsite.37 As observed with compounds **1a** and **1b**, a strong difference in efficiency was observed for both diastereomers. These data strongly suggest that 7b mainly binds cyclophilin via an interaction at the S1' subsite. Therefore, the $Gly\psi(PO_2Et-N)$ -Pro moiety plays a major role in the interaction. Even though the affinity of compound **1b** is modest compared to those of nanomolar inhibitors of proteases, the results reported herein suggest that the phosphonamide-proline pattern might be a transition-state mimic of the PPIase activity of hCyp-18.

To compare between phosphonamide and ketoamide, we tested peptide 1**3** which contains a glycine analogue of the ketoamide motif $Ala\psi(COCO-N)$ Pro, a canonical transition-state mimic of PPIases.²⁵ We were surprised to notice that compound **13** inhibits hCyp-18 with a relatively low efficiency (Table 2). Several reasons may be given for this lack of potency: (i) the conformation of the Gly $\psi(COCO-N)$ Pro moiety, related to the particular behavior of the Gly-Pro moiety compared to other Xaa-Pro motifs,²⁰ and (ii) a putative tendency of the aminopyruvyl motif to enolize, facilitated by the absence of electron-donating groups at C3 of the aminopyruvate.

In conclusion, we have reported the design, synthesis, and biochemical evaluation of $Gly\psi(PO_2Et-N)Pro$ containing peptides as novel transition-state analogues of the PPIase activity of hCyp18. The biochemical assays have shown that pseudopeptide **1b** is a good and selective inhibitor of cyclophilin, whereas the $Gly\psi$ -

Gly ψ *(PO₂R-N)Pro-Containing Pseudopeptides*

(COCO-N)Pro equivalent has only a moderate affinity for the enzyme. This suggests that the phosphonamide motif is an interesting mimic of the rotating deconjugated amino acyl-proline amide bond. These results open the way to the development of more potent, selective inhibitors of hCyp-18.

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Supporting Information Available: Experimental procedures and characterizations of pseudopeptides 7-13. This material is available free of charge via the Internet at http:// pubs.acs.org.

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